# Introduction to Physical Science

Temperature Presented by Robert Wagner

#### Temperature

- Temperature to humans can be a matter of perception.
  - If you place one hand in hot water and another hand in cold water. What happens when you place both hands in room temperature water?
- Scientifically, temperature is a measure of the average velocities of the particles in a substance
  - · This is what is measured with a thermometer

#### Linear Thermal Expansion

- The change in length of a substance depends on the temperature, material and the length itself:
  - .
  - is the coefficient of linear expansion
- · Coefficient of linear expansion
  - Depends on the properties of the materials involved

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Material	Coefficient of linear expansion a(1/*C)	Coefficient of volume expansion g(1/°C)
Solids		
Aluminum	$25 \times 10^{-6}$	$75 \times 10^{-6}$
Brass	$19 \times 10^{-6}$	56 × 10 <sup>-6</sup>
Copper	17×10 <sup>-6</sup>	51×10 <sup>-6</sup>
Gold	$14 \times 10^{-6}$	$42 \times 10^{-6}$
Iron or Steel	$12\times10^{-9}$	$35 \times 10^{-6}$
Invar (Nickel-iron alloy)	$0.9 \times 10^{-6}$	2.7 × 10 <sup>-6</sup>
Lead	$29 \times 10^{-6}$	$87 \times 10^{-6}$
Silver	$18 \times 10^{-6}$	54 × 10 <sup>-6</sup>
Glass (ordinary)	$9\times10^{-6}$	$27 \times 10^{-6}$
Glass (Pyreoth)	3×10 <sup>-6</sup>	9×10 <sup>-6</sup>
Quartz	$0.4 \times 10^{-6}$	$1 \times 10^{-6}$
Concrete, Brick	$-12 \times 10^{-6}$	$-36 \times 10^{-6}$
Marble (average)	7×10 <sup>-6</sup>	$2.1 \times 10^{-5}$
Liquids		
Ether		$1650 \times 10^{-6}$
Ethyl alcohol		$1100 \times 10^{-6}$
Petrol		$950 \times 10^{-6}$
Glycerin		$500\times10^{-6}$
Mercury		$180 \times 10^{-6}$
Water		210×10 <sup>-6</sup>
Gazes		
Air and most other gases at atmospheric pressure		$3400 \times 10^{-6}$

#### Example

- Room temperature is generally considered to be about . What would this be in ? In Kelvins?
  - · Draw a sketch
  - Identify known values
  - Identify equation
  - · Enter values in the equation and solve

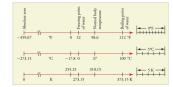
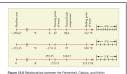


Figure 13.6 Relationships between the Fahrenheit, Celsius, and Kelvin temperature scales, rounded to the nearest degree. The relative sizes of the

Image Credit: OpenStax College Physics - Figure 13.6 CC BY 4.0

#### Example

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$$T(^{o}C) = 25.0^{o}C$$

$$T(^{o}F) = \frac{9}{5}T(^{o}C) + 32$$

$$T(^{o}F) = \frac{9}{5}(25.0) + 32 = 77.0^{o}F$$

$$T(K) = T({}^{o}C) + 273.15$$

$$T(K) = 25.0 + 273.15 = 298K$$

Image Credit: OpenStax College Physics - Figure 13.6 CC BY 4.0

#### Thermal Equilibrium

- If two systems, A & B are in thermal equilibrium with each other, and B is in thermal equilibrium with a third system, C, then A is also in thermal equilibrium with C.
  - The Zeroth Law of Thermodynamics
- · Heat will flow from hotter object to cooler objects, equalizing their temperatures.
  - This is how a thermometer measures temperature

Image Credit: OpenStax College Physics - Figure 8.8

#### Thermal Expansion

- Thermal expansion is related to the change in temperature
  - Greater temperature change means greater expansion
- · The amount of expansion also depends on the material
  - · Alcohol in a thermometer expands more than the glass



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#### Example

- The span of a bridge is 1275 m long at its coldest. The bridge is exposed to temperatures ranging from
- . What is the change in length between these temperatures if the bridge is made of steel?
- · Draw a sketch
- · Identify known values
- · Identify equation
- · Enter values in the equation and solve



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#### Example

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$$L=1275~m~;~\Delta T=55^{o}C~;~\alpha(steel)=\frac{12x10^{-6}}{^{o}C}$$

$$\Delta L = \alpha L \Delta T$$

$$\Delta L = (\frac{12x10^{-6}}{{}^{o}C})(1275 \ m)(55^{o}C)$$

$$\Delta L = 0.84 \, m$$

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## Thermal Expansion in Two and Three Dimensions

- · Two Dimensions:
  - •
- · Three Dimensions
  - •
- is the coefficient of volume expansion

#### Example

- Suppose a 60.0 L steel gasoline tank is full. The gas and tank have a temperature of 15.0°C. How much gasoline will spill by the time they warm to 35.0°C?
  - · Draw a sketch
  - · Identify known values
  - · Identify equation
  - · Enter values in the equation and solve

$$V = 60.0 L; \ \Delta T = 20.0^{\circ}C; \ \beta(steel) = \frac{35x10^{-6}}{{}^{\circ}C}$$
$$\beta(gas) = \frac{950x10^{-6}}{{}^{\circ}C}$$

$$\Delta V_s = \beta_s \Delta T$$
;  $\Delta V_{eas} = \beta_{eas} V_{eas} \Delta T$ 

$$V_{spill} = \Delta V_{gas} - \Delta V_s$$
;  $V_{gas} = V_s$ 

$$V_{spill} = (\beta_{gas} - \beta_{s})V\Delta T$$

$$V_{spill} = \frac{(950 - 35)x10^{-6}}{{}^{o}C}(60.0 L)(20.0 {}^{o}C) = 1.10 L$$

#### Thermal Stress

- Thermal stress is caused by expansion and contraction as the temperature changes.
  - · Damage to roads potholes
  - · Weathering of rocks
  - Rupturing of tank
  - · Glass cooking pans
  - · Dental fillings



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### Summary

- Temperature is a measure of the average kinetic energies of the particle in a substance
- Objects in contact will reach thermal equilibrium as heat is transferred from a hotter object to a cooler one
- Thermal stress occurs when changes in temperature are rapid or when material in unable to expand and contract freely